

SESQUITERPENES FROM *RUDBECKIA GRANDIFLORA*

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Key Word Index—*Rudbeckia grandiflora*; Asteraceae; Heliantheae; sesquiterpene esters; eudesmanes; sesquiterpene lactones; pseudoguaianolides.

Abstract—Chemical analysis of *Rudbeckia grandiflora* afforded two new sesquiterpene esters, 6 α -hydroxycostic acid, methyl ester and arbusculin E, methyl ester as well as two new pseudoguaianolide type sesquiterpene lactones, desacylligulatin C and rudbeckin A. Their structures were elucidated by spectroscopic methods and chemical transformations.

INTRODUCTION

The genus *Rudbeckia* L. consists of ca 18 species and 30 taxa of annual and perennial herbs divided into two subgenera, *Rudbeckia* and *Macrocline*. *Rudbeckia grandiflora* is very abundant along roadsides and in pastures in Arkansas, Oklahoma and Louisiana. Previous chemical studies of members of this genus included *R. mollis* [1] and *R. laciniata* [2]. Sesquiterpene lactones isolated from *R. mollis* exhibited antitumour-activity [1].

As part of a biochemical systematic study within the tribe Heliantheae combined with a search for bioactive plant products we have analysed the previously uninvestigated species *R. grandiflora*. The structure of the four new compounds which we named 6 α -hydroxycostic acid methyl ester (1), arbusculin E methyl ester (2), desacylligulatin C (3) and rudbeckin A (4), was established by chemical and spectroscopic methods.

RESULTS AND DISCUSSION

The dichloromethane extract of floral parts of *R. grandiflora* afforded, after column chromatography and further purification by preparative TLC, four new compounds. 6 α -Hydroxycostic acid methyl ester (1), C₁₆H₂₄O₃, exhibited in the ¹H NMR spectrum a three-proton singlet at δ 0.82 suggesting an angular methyl in a eudesmane skeleton. Two one-proton broadened singlets at δ 6.28 and 5.22 together with a three-proton singlet at δ 3.79 indicated an α -methylene methyl ester of the costic acid type [3]. Two broadened singlets at δ 5.00 and 4.67 were in agreement with an olefinic methylene group which had to be on C-15 on biogenetic grounds. A doublet of doublets at δ 3.92 was assigned to a proton on a carbon bearing a hydroxyl group which showed a coupling to H-5, a doublet at δ 1.91 (J = 10.1 Hz), and to H-7 (ddd , δ 2.60) as determined by 2D COSY exper-

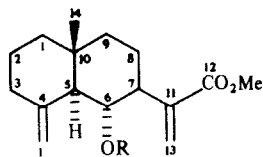
iments. This finding unambiguously established the position of the hydroxyl group at C-6. The stereochemistry at C-6 was derived from the couplings between H-5, H-6 and H-7 ($J_{5,6} = J_{6,7} = 10.1$ Hz) which were in agreement with a *trans*-diaxial orientation of these three protons. Assuming an H-7 α as in all sesquiterpenes from higher plants [4] the hydrogens at C-5 and C-6 had to be α and β , respectively. This was further substantiated by NOE difference experiments which showed effects between H-5 and H-7 but had no effect on H-6. *In situ* acylation of the hydroxyl group with trichloroacetyl isocyanate (TAI) [5] corroborated the above findings. The ¹H NMR spectrum of the trichloroacetylcarbamate (TAC) derivative (1a) showed one NH signal at δ 8.20, providing further evidence for the presence of one hydroxyl group in compound 1. The paramagnetic acylation shift of the doublet of doublet at δ 3.92 (H-6) in 1 to δ 5.5 in 1a ($\Delta\delta$ = 1.58) was consistent with a geminal position of H-6 and the hydroxyl group. The signal of the C-10 methyl group at δ 0.82 showed no paramagnetic shift, implying that the C-10 β -methyl is oriented away from H-6, which is in agreement with the relative stereochemistry of the two groups, that is, C-6 α -hydroxyl and C-10 β -methyl.

All other proton absorptions of ester 1 were assigned on the basis of 2D COSY, spin decoupling and NOE difference experiments (Table 1). The ¹³C NMR spectrum of 1 (Table 2) was assigned with the aid of the heteronuclear multipulse DEPT experiments; the data of this new sesquiterpene supported the ¹H NMR assignments of 1, which was named 6 α -hydroxycostic acid methyl ester.

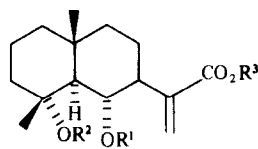
Arbusculin E methyl ester (2), C₁₆H₂₄O₄. The ¹H NMR spectrum (Table 1) values are very similar to those of 6 α -hydroxycostic acid methyl ester (1), the major differences between 1 and 2 residing in the proton signals of C-4/C-15. Instead of the two olefinic protons at C-15 in 1 compound 2 exhibited a three-protons singlet at δ 1.32, which is characteristic of a methyl group on a carbon bearing a hydroxyl group. The chemical shift of H-5 in 2 was also shifted upfield by about 0.5 ppm suggesting that it was in a non-allylic position. Therefore, the methylene group at C-4 in 1 had to be changed to a methyl group at a carbon bearing a hydroxyl group in 2. The ¹³C NMR spectrum of 2 (Table 2) corroborated the proposed structure with signals at δ 106.92 (C-15) and

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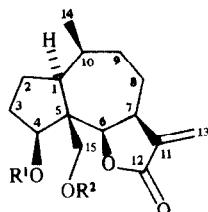
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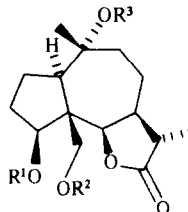
1 R = H
1a R = CONHCOCCl₃



2 R¹ = R² = H, R³ = Me
2a R¹ = R² = R³ = H
2b R¹ = R² = CONHCOCCl₃, R³ = Me



3 R¹ = R² = H
3a R¹ = R² = CONHCOCCl₃



4 R¹ = R² = R³ = H
4a R¹ = R² = R³ = CONHCOCCl₃

14.3 (C-4) in the spectrum of **1** being replaced by signals at δ 23.78 (C-15) and 73.47 (C-4) in compound **2**.

The stereochemistry at the chiral centres was proposed on the basis of the results of the TAI experiments and by examination of stereo models. The ¹H NMR spectrum of the trichloroacetylcarbamate **2b** showed NH signals at δ 8.20 and 8.30 indicating two hydroxyl groups in compound **2**. The signal due to H-6 at δ 4.10 in the spectrum of compound **2** was shifted to δ 5.28 in that of **2b** ($\Delta\delta$ = 1.18) due to the strong deshielding effect of the TAC acyl group. Also, the C-4 methyl group was shifted from δ 1.32 in the spectrum of **2** to δ 1.60 in that of **2b** ($\Delta\delta$ = 0.28). These shifts strongly suggested an α -orientation of the two hydroxy groups in the molecule as well as the α -orientation of H-5. The structure of **2** was unambiguously established by its transformation to 6 α -hydroxycostic acid (**2a**) with NMR spectral data being in complete agreement with data reported in the literature [3]. Compound **2** which was not previously reported as a natural product, represents the methyl ester of the known arbusculin E [7].

Desacylligulatin C (**3**), C₁₅H₂₂O₄, is a gum with ¹H NMR spectral signals typical for an α -methylene- γ -lactone moiety with two one-proton doublets at δ 6.20 (H-13b) and 5.49 (H-13a), both coupled to a one-proton multiplet at δ 3.45 (H-7). Strong IR absorptions at 1757 and 3429 cm⁻¹ corroborated the presence of a γ -lactone and hydroxyl(s), respectively. 2D COSY studies indicated that H-7 was coupled to a doublet at δ 4.51 ($J_{6,7}$ = 9.5 Hz) which was assigned to the lactonic proton at C-6. The multiplicity of H-6 suggested a pseudoguaianolide-type skeleton for lactone **3** [4]. A three-proton doublet at δ 0.99 (J = 7.5 Hz) was assigned to the methyl group at C-10. However, the common methyl signal due to an angular methyl at C-5 was missing in compound **3**. Instead, a pair of geminally coupled doublets (J = 12.2 Hz) at δ 4.05 and 3.90 suggested the presence of a -CH₂OH group at C-5, which was supported by a ¹³C NMR triplet at δ 61.82. Compound **3** lacked a ¹³C NMR absorption for a

cyclopentanone carbonyl but exhibited a doublet at δ 83.91 typical of a carbon bearing an oxygen. The ¹H NMR spectrum supported the above findings by the presence of a doublet of a doublet at δ 4.35 (J = 8.8 and 9.0 Hz). The chemical shift together with the multiplicity of this proton suggested the presence of a hydroxyl group at C-4. All other proton signals were assigned on the basis of spin decoupling experiments, 2D COSY and 2D ¹H-¹³C correlations (Table 1). The ¹³C NMR spectral signals (Table 2) were assigned on the basis of 2D COSY, ¹H-¹³C correlation and DEPT experiments. The spectroscopic data were consistent with data reported for a synthetic derivative of ligulatin C [6]. To our best knowledge this is the first report of desacylligulatin C (**3**) as a natural product.

Rudbeckin A (**4**), C₁₅H₂₄O₅, showed an IR band at 1765 cm⁻¹ (γ -lactone) and the ¹H NMR (Table 1) also suggested a lactone which had to be closely related to desacylligulatin C (**3**). Compound **4** differed from **3** in that the two doublets characteristic of an α -methylene- γ -lactone were missing in the spectrum of compound **4**. Instead, a three-proton doublet at δ 1.35 (J = 7.8 Hz) appeared which was assigned to the C-11 methyl group. The doublet at δ 0.99 (C-10-Me) in the spectrum of **3** was replaced by a singlet at δ 1.3 characteristic of a methyl group at a carbon bearing a hydroxyl and H-7 at δ 2.25 appeared to be non-allylic. The ¹³C NMR spectrum (Table 2) was assigned by comparison of **4** with lactone **3**. Application of heteronuclear multipulse DEPT experiments and 2D COSY allowed assignment of all ¹H NMR spectral signals.

The above spectral data were in agreement with structure **4** exclusive of its stereochemistry, which was determined with the aid of NOE difference spectral data and TAI experiments. Irradiations of the C-11 methyl signal from a NOE difference experiment showed an effect on H-6 and H-7, indicating an α -orientation of the C-11 methyl group. The ¹H NMR spectrum of the TAC derivative (**4a**) exhibited paramagnetic acylation shifts of H-

Table 1. ^1H NMR spectral data of compounds 1–4 (400 MHz, CDCl_3 , TMS as internal standard)*

1			2			3			4	
H	CDCl ₃	C ₆ D ₆ †	CDCl ₃	C ₆ D ₆ †	CDCl ₃	C ₆ D ₆ †	CDCl ₃		CDCl ₃	
1α	1.58–1.66	m‡	1.51–1.59‡	1.70	ddd	1.55–1.46‡	1.65	ddd	2.05	dd
1β	1.38	ddd	1.10	1.40	ddd	1.33	—	—	—	—
2α	1.48	dddd	1.24	1.51	dddd	0.99	1.51	dddd	1.55	dddd
2β	1.58–1.66	m‡	1.45	1.38	m‡	1.10	1.76–1.92	m‡	1.91–1.97	m‡
3α	2.00	ddd	1.77	2.26	ddd	1.55–1.46‡	1.76–1.92	m‡	1.89	dddd
3β	2.34	ddd	2.15	2.01	ddd	1.75	2.20	dddd	1.87	dddd
4	—	—	—	—	—	—	4.35	dd	4.24	dd
5α	1.91	d	1.67	1.45	d	1.37	—	—	—	—
6	3.92	dd	4.00	4.10	dd	4.16	4.51	d	4.75	d
7	2.60	ddd	2.70	2.60	ddd	2.63	3.45	ddd	2.25	ddd
8α	1.58–1.66	m‡	1.51–1.59‡	1.78	dddd	1.68	1.90	dddd	1.54	dddd
8β	1.72	dddd	1.68	1.61	dddd	1.47	2.09	dddd	2.10	dddd
9α	1.41	ddd	1.17	1.73	ddd	1.08	1.58	ddd	1.68	ddd
9β	1.32	ddd	1.05	1.79	ddd	1.72	1.75	dddd	1.91–1.97	m‡
10	—	—	—	—	—	—	2.02	ddq	—	—
11β	—	—	—	—	—	—	—	—	2.35	dq
13b	6.28	br s	6.30	6.28	br s	6.20	6.20	d	1.35	d
13a	5.22	br s	5.48	5.22	br s	5.45	5.49	d	—	—
14	0.82	s	0.78	0.98	s	0.79	0.99	d	1.27	s
15	5.00	br s	4.82	1.32	s	1.31	4.05	d	4.17	d
15′	4.67	br s	4.62	—	—	—	3.90	d	4.05	d
OMe	3.79	s	3.4	3.79	s	3.38	—	—	—	—

Compound 1 $J(\text{Hz})$: 1 α ,2 α = 1 β ,2 α = 7,8 α = 8 β ,9 β = 3.5; 1 β ,1 α = 2 α ,2 β = 8 α ,8 β = 13.1; 1 β ,2 β = 4.6; 2 α ,3 α = 8 α ,9 β = 6.4; 2 α ,3 β = 1.9; 3 β ,2 β = 8 α ,9 α = 4.0; 2 β ,3 α = 9 α ,9 β = 12.0; 3 α ,3 β = 12.8; 5,6 = 6,7 = 10.1; 7,8 β = 9; 8 β ,9 α = 11. Compound 2 $J(\text{Hz})$: 1 α ,2 α = 3.5; 1 α ,2 β = 2.30; 1 α ,1 β = 12; 1 β ,2 β = 1 β ,2 α = 7,8 α = 2 α ,3 α = 3.30; 2 α ,2 β = 17; 2 β ,3 α = 12; 3 α ,3 β = 9.7; 3 β ,2 β = 2 α ,3 β = 8 α ,9 α = 4.0; 5,6 = 6,5 = 6,7 = 7,8 β = 10.4; 7,13 = 0.8; 8 α ,8 β = 5; 8 α ,9 β = 5.8. Compound 3 $J(\text{Hz})$: 1,10 = 8.9; 1,2 α = 10.5; 1,2 β = 5.9; 2 α ,2 β = 12.5; 2 α ,3 β = 8.9; 2 α ,3 α = 4.5; 3 β ,4 = 9.0; 3 α ,3 β = 14.3; 3 β ,2 β = 6.0; 4,3 β = 9.0; 4,3 α = 8.8; 6,7 = 9.5; 7,8 β = 7,8 α = 8 α ,9 β = 13.3; 8 α ,9 α = 3; 8 α ,8 β = 17.8; 8 β ,9 α = 8 β ,9 β = 7.4; 9 α ,9 β = 11.8; 9 α ,10 = 0.0; 9 β ,10 = 11; 10,14 = 7.5; 13a,7 = 3.8; 13b,7 = 3.4; 15,15' = 12.2. Compound 4 $J(\text{Hz})$: 1,2 α = 4.5; 1,2 β = 10; 2 α ,2 β = 12.5; 2 α ,3 α = 4.5; 2 α ,3 β = 10.5; 3 α ,3 β = 12.6; 3 α ,2 β = 3 α ,4 = 7.8; 3 β ,4 = 5.6; 3 β ,2 β = 6.0; 6,7 = 6.7; 7,8 α = 0.0; 7,11 = 1.8; 7,8 β = 12.7; 8 α ,9 α = 3; 8 α ,8 β = 17.8; 8 α ,9 β = 13.3; 8 β ,9 α = 8 β ,9 β = 9 α ,8 β = 7.4; 9 α ,8 α = 3; 9 α ,9 β = 11.8; 9 α ,10 = 0.0; 9 β ,8 α = 13.3; 9 β ,10 = 11; 13,11 = 3.8; 15,15' = 12.2.

*The signals due to hydroxyl groups are omitted.

\ddagger multiplicity and coupling constants are the same.

\ddagger Obscured by other signals.

4 ($\Delta\delta$ = 1.16), H-15 ($\Delta\delta$ = 1.35) and H-1 ($\Delta\delta$ = 0.20) establishing the presence of hydroxyl groups at C-4 and C-15 and the α -orientation for the hydroxyl group at C-10. An axial β -hydroxyl group should have caused a noticeable shift of the H-9 α and H-2 β signals, which were not observed. The C-11 methyl signal was not shifted significantly which is consistent with an α -orientation for the C-11 methyl group. Similarly, the H-6 α signal was not affected suggesting a β -orientation of the CH_2OH group at C-5. This new pseudoguaianolide has not been previously reported and is named rudbeckin A (4).

EXPERIMENTAL

Rudbeckia grandiflora (Sweet) DC. was collected on 7 June 1986, in pinewoods and roadside of Louisiana Hwy 121, 3.3 miles west of Rapides Parish line east of La Camp. sec 16, T 2N, R 5W. (P. Cox, L. E. Urbatsch and E. Harris; No. Cox 4889, voucher deposited at L.S.U., U.S.A.).

The air-dried flowers (50 g) were ground and extracted according to the general procedure [8] providing 1.3 g of the

crude terpenoid extract. The crude extract (1.0 g) was separated by CC on silica gel using EtOAc–hexane mixtures of increasing polarity, 86 fractions of 50 ml each being collected. Upon further prep. TLC of the various fractions two sesquiterpenes (1 and 2) and two sesquiterpene lactones (3 and 4) were isolated. Fraction 12 provided 30 mg of 1 and fractions 24, 25 gave 34 mg of 2. Fractions 29–31 afforded 3 (50 mg) and fractions 40, 41 provided 4 (35 mg).

6 α -Hydroxycostic acid methyl ester (1). $\text{C}_{16}\text{H}_{24}\text{O}_3$, colourless gum; IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3511 (OH), 1719 (ester), 1647 and 1626 (double bond); HRMS (FAB, probe) 70 eV, m/z (rel. int.): 264.1247 [M] $^+$ (4) (calc. for $\text{C}_{16}\text{H}_{24}\text{O}_3$: 264.1249), 249.1330 [M – Me] $^+$ (2), 233.1145 [M – OMe] $^+$ (2), 186.0713 (64), 185.0781 (96), 93.0417 (100); ^1H NMR see Table 1; ^{13}C NMR see Table 2; TAC-derivative of 1: ^1H NMR of H-6, δ 3.92 ($\Delta\delta$ = +1.58).

Arbusculin E methyl ester (2). $\text{C}_{16}\text{H}_{24}\text{O}_4$, colourless gum; IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3347 (OH), 1719 (ester) 1626 and 1439 (double bond); HRMS (FAB, probe) 70 eV, m/z : 282.1541 [M] $^+$ (37) (calc. for $\text{C}_{16}\text{H}_{24}\text{O}_4$: 282.1544), 265.1917 (75), 251.1692 [M – OMe] $^+$ (8), 247.1750 (37), 233.1542 (74), 161.1285 (48), 133.8546 (100), 109.0871 (45); ^1H NMR see Table 1; ^{13}C NMR see Table 2; TAC-derivative of 2: ^1H NMR: H-6, δ 4.10 ($\Delta\delta$ = +1.18); H-4, δ 1.32 ($\Delta\delta$ = +0.28).

Table 2. ^{13}C NMR data of compounds 1–4 (50.32 MHz, CDCl_3 , TMS as internal standard)*

C	1	2	3†	4†
1	41.86t‡	42.92t‡	48.45d	46.89d
2	26.66t	19.53t	22.30t	22.69t
3	37.82t	43.59t	29.70t	27.16t
4	14.30s	73.47s	83.91d	82.74d
5	57.92d	57.92d	53.52s	60.23s
6	69.33d	73.26d	90.09d	85.71d
7	48.07d	50.25d	43.44d	43.99d
8	23.93t	26.80t	32.29t	31.86t
9	40.38t‡	42.63t‡	24.80t	42.41t
10	37.40s	36.31s	32.76d	85.31s
11	147.27s	142.27s	139.88s	53.00d
12	168.07s	168.33s	170.23s	179.45s
13	124.63t	125.75t	118.89t	16.64q
14	17.61q	19.70q	13.95q	24.32q
15	106.92t	23.78q	61.82t	68.91t
OMe	51.84q	52.01q	—	—

*Peak multiplicity was obtained by heteronuclear multipulse programs.

†Assignments for compounds 3 and 4 were confirmed by ^{13}C – ^1H chemical shift correlation.

‡Assignments may be interchanged.

Saponification of compound 2. A 10 mg sample of 2 was dissolved in 3 ml of 5% methanolic KOH. After 16 hr the soln was diluted with H_2O (10 ml) and neutralized with 1% HCl. Extraction with DCM gave 5 mg of 2a which was purified by prep. TLC [3].

Desacylligulatin C (3). $\text{C}_{15}\text{H}_{22}\text{O}_4$, gum; IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3429 (OH), 1757 (γ -lactone), 1663 (double bond). HRMS (FAB, probe) 70 eV m/z : 266.1436 $[\text{M}]^+$ (3), (calc. for $\text{C}_{15}\text{H}_{22}\text{O}_4$: 266.1434), 248.1377 $[\text{M}-\text{H}_2\text{O}]^+$ (6) 235.1268 $[\text{M}-\text{CH}_2\text{OH}]^+$ (19), 230.1105 $[\text{M}-2\text{H}_2\text{O}]^+$ (17), 203.1301 (68), 159.1097 (100),

147.0913 (83), 145.0772 (94), 105.0594 (88), 95.0637 (84). ^1H NMR see Table 1; ^{13}C NMR see Table 2.

TAC-derivative of 3: ^1H NMR: H-15', δ 3.90 ($\Delta\delta = +0.62$), H-15, δ 4.05 ($\Delta\delta = +0.55$), H-4, δ 4.35 ($\Delta\delta = +0.69$), H-6, δ 4.51 ($\Delta\delta = +0.99$).

Rudbeckin A (4). $\text{C}_{15}\text{H}_{24}\text{O}_5$, gum; IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{-1} : 3418 (OH), 1765 (γ -lactone); HRMS (FAB, probe) 70 eV m/z 266.1614 $[\text{M}-\text{H}_2\text{O}]^+$ (26) (calc. for $\text{C}_{15}\text{H}_{22}\text{O}_4$: 266.1616), 265.1161 (76), 251.1371 (74), 249.1264 (82), 231.1502 $[\text{M}-2\text{H}_2\text{O}]^+$ (30), 219.1437 $[\text{M}-3\text{H}_2\text{O}]^+$ (31), 203.1137 (62), 185.0731 (94), 93.0718 (100); ^1H NMR see Table 1. ^{13}C NMR see Table 2.

TAC-derivative of 4: ^1H NMR: H-4, δ 4.24 ($\Delta\delta = +1.16$); H-15, δ 4.05 ($\Delta\delta = +1.35$), H-1, δ 2.25 ($\Delta\delta = +0.20$).

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